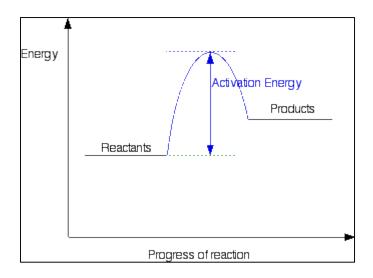
MINISTRY OF EDUCATION, HERITAGE AND ARTS YEAR 12 CHEMISTRY REVISION WORKSHEET 7

Write the answers to the following questions in your exercise/activity books.

Strand 3: Reactions

1. Study the diagram given below and answer the questions that follow.



Source: https://www.chemguide.co.uk

(i). State if the energy profile diagram shown above is of an endothermic or exothermic reaction.

(1 mark)

(ii). Explain your answer in part (i) above.

(2 marks)

2. The reaction between zinc metal and hydrochloric acid solution gives out heat.

$$Zn_{(s)} \quad + \quad 2HCl_{(aq)} \quad \rightarrow \quad ZnCl_{2(aq)} \quad + \quad H_{2(g)}$$

(i). State the sign of the $\triangle_r H$ for the above reaction?

(1 mark)

(ii). Sketch an energy diagram for the reaction.

Note: Label the reactants and products on the diagram.

(4 marks)

THE END